MANY-BODY METHODS FOR ATOMS AND MOLECULES

Rajat Kumar Chaudhuri Sudip Kumar Chattopadhyay



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Preface

The development of quantum mechanics in the 1920s, coupled with the genesis of advanced computational algorithms and techniques, culminated in a solid foundation for the development of quantum chemistry, aimed at solving problems associated with atomic and molecular systems, in a wide range of contexts. The problem becomes non-trivial since the many-body Schrödinger equation is not solvable exactly even for the smallest of systems, and one has to resort to techniques of approximation. The prime focus of quantum chemists is toward the development of accurate, formally simple, computationally tractable, and consistently applicable methods of approximation that can be applied to a plethora of systems having arbitrary levels of complexity and generality.

Thus, electronic structure theorists are confronted with the challenge of dealing with the electron correlation problem for many-particle systems. The present book deals with two major classes of theories of electron correlation, namely, the many-body perturbation theory and coupled cluster methods. In the parlance of modern strategies dealing with electronic structure, both methods have withstood the test of time and proved to be of great importance in chemical physics. While the linked cluster theorem takes care of single reference situations, both in the context of coupled cluster and perturbation theories, the corresponding multireference analog is yet to be realized with full-fledged generality. Multireference situations appear in many cases in the domain of chemistry and physics, in general, where the traditionally accepted single reference techniques seem to be inadequate, or at times formally wrong, and often lead to an incomplete physical insight. In view of this, recent decades have witnessed a tremendous methodological development in the realm of multireference electronic structure methods, both in coupled cluster and perturbative contexts, and have created a profound impression in the arena of modern electronic structure theories. Such a set of multireference developments in the realm of perturbation theories sets the tone of the present book.

This book primarily aims to discuss the issues related to the formal development and the consequent numerical implementation of the multireference coupled cluster and perturbation theories from the standpoint of a practicing theoretician. The theoretical nitty-gritty associated with such developments has been emphasized amply, and the related algebraic (and many times diagrammatic) derivations have been put forth in a comprehensive manner to

Preface

assist the reader to have a grip on the subject. It has been assumed, throughout this work, that the reader is acquainted with the fundamental nuances of quantum chemistry. There are, nevertheless, numerous textbooks dealing with quantum chemistry at the beginner's level, like the one by Szabo and Ostlund, and a good number of professionally written monographs, like the one by Shavitt and Bartlett [Many-Body Methods in Chemistry and Physics: MBPT and Coupled Cluster Theory] that pave the foundation to the modern nuances of electronic structure. This book is more like an advanced reference text, that would carry forward the reader from the doorstep of modern research to the very frontier of it.

The book is organized as follows. The opening chapter, Chapter 1, attempts to introduce the reader, in a very succinct manner, to a set of issues relevant to the development of explicitly correlated many-body methods in general. We initiate this chapter by formally dealing with the basic issues like the Born–Oppenheimer approximation and general treatment of approximation methods, and proceed to discuss atomic/molecular electronic structure methods by setting the tone with the trivial-most independent particle model. We move ahead, in a very concise fashion, to discuss advanced methods like configuration interaction. As we have already pointed out, since this book is not a beginner's guide to modern electronic structure and is attuned more toward advanced readers, we have given extensive bibliographical support to augment the issues that we have left out of this book.

Chapter 2 is a formally rigorous treatment of the issues that pave the foundation toward the theoretical development of methods attuned to tackle the problem of electronic correlation in a balanced and systematic manner. This chapter is somewhat technical in the sense that here we have developed the pivotally crucial hole-particle formalism through the normal ordering of the creation-annihilation operators using Wick's theorem. In addition, we have also introduced the diagrammatic representation of the involved operator expansions, which eventually leads to the diagrammatic simplification of the working equations (in the chapters that follow), as compared to their complete algebraic form.

In Chapter 3 we consider the perturbation theory, initially in the philosophy of a typical textbook pattern to set the tone of the development. As we progress, we deal with the projection operator route to the perturbation theory, and the Rayleigh–Schrödinger and the Brillouin–Wigner variants in the realm of a completely single-reference situation. In doing so, we discuss in detail the underlying issues that lead to intuitively intelligent partitioning of the Hamiltonian, and have ultimately compared and contrasted the relative performance of the variants under different sets of conditions.

We carry forward the ideas developed in Chapter 3 to deal with perturbation theory in the multireference domain in Chapter 4 by invoking the idea of an appropriate multiconfiguration space and suitably redefine the orbital space as being tailored into core, active, and valence. We bring to the fore the issue of improved virtual orbitals and discuss the formal issues associated with multireference perturbation theories in both complete and incomplete model spaces.

In Chapter 5 we formally introduce the idea of state-specific theories by discussing the second-order state-specific multireference perturbation theory in the context of a Möller–Plesset partitioning. Various other multiconfiguration and multistate variants are also addressed in this chapter.

Chapter 6 is dedicated to the development and application of the traditional coupled cluster techniques in the single-reference context. In this chapter, we address the crucial issue of the size-extensive nature of the computed energies, and compare the coupled cluster method with the traditional configuration interaction technique. We put forth a detailed survey of the methods to represent the working equations in a diagrammatic format, leading to the extensive simplification of the algebraic forms involved. We also show the method to develop perturbation theories starting from a coupled cluster equation. Numerical applications are provided to supplement the theoretical discussion in this chapter.

Chapter 7 deals with Fock-space based multireference coupled cluster methods. In this chapter we delineate the issues of the choice of the wave operator for multireference systems, the connectivity of the effective Hamiltonian, and the systematic generation of cluster equations for various valence sectors. We deal with the equation of motion coupled cluster method, and its relation to the Fock-space-based multireference coupled cluster technique. Numerical illustrations form a major attraction of this chapter, since these provide a suitable window for the reader to judge the efficacy of the different methods discussed here and the preceding ones.

We conclude this book with Chapter 8, which deals with *state-of-the-art* methods of modern electronic structure. We emphasize the state-universal and the state-specific methods in the contexts of perturbation and coupled cluster theories, and assess their relative performances. This chapter discusses the frontier areas of explicitly correlated methods and provides the reader with a scope to push the frontier ahead.

We want to thank all the people who have made it possible for this book to reach your hands. First, we are deeply indebted to Professor Debashis Mukherjee and Professor Karl F. Freed for the training that they bestowed upon us. We are thankful to Professor A. Satyanarayan and Professor R. Ramesh for their advice and constructive criticism on the entire manuscript. Finally, we express our deep sense of gratitude and thankfulness to the editorial team of Taylor & Francis for providing us this opportunity.



Chapter 1

Introduction

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1.1 Background

Perturbation theory, in the arena of quantum mechanics, refers to a set of physically motivated and intuitive approximations engineered by mathematical perturbation to describe and explore the properties of a complicated system. The philosophy of perturbation theory is to begin with a simpler system, whose mathematical solution is known to the best of one's knowledge, and then "perturb" the system by means of an externally tunable influence. For a disturbance that is not too strong, the physical quantities associated with the perturbed system can be expressed in terms of the "corrections" of the corresponding quantities of the simpler (unperturbed) system. For small perturbation, the magnitudes of these correction terms are really small in light of the magnitude of the quantities themselves and are therefore subject to treatments by approximations like the asymptotic series. The accuracy of the evaluated physical quantity, among other factors, depends on the extent to which such correction terms are incorporated. Therefore, perturbation theory is a useful avenue to explore the properties of a complicated system, where the exact (analytic) solution of the system is either difficult or impossible to achieve.

In early 1950s, Löwdin [1, 2, 3] and others [4] demonstrated that the "partitioning technique" can be used to develop perturbation theory for manyparticle systems. A series of seminal works by P. O. Löwdin [5] entitled "Studies in Perturbation Theory" during 1963–1971 led to a deep insightful understanding of the expansions of perturbation theory. These explorations also provided pivotal knowledge of the interrelations among the various forms of such expansions, and their immediate consequence and capabilities in tackling quantum mechanical problems pertaining to chemistry.

The perturbation theory for many-electron closed-shell systems [the socalled many-body perturbation theory (MBPT)] was formally developed during the 1950s by Brueckner [6, 7], Goldstone [8], Hugenholtz [9] and Hubbard [10]. Kelly [11] was probably the first to apply this method to the atomic structure problem. These applications used the numerical solutions of the Hartree-Fock equations, which are available for atoms, because of the special coordinate system. Kelly also reported applications to some simple hydrides in which the hydrogen atom nucleus is treated as an additional perturbation. Historically, the diagrammatic representation of MBPT was first introduced by Brueckner [6, 7] and by Brueckner and Levinson [12] in nuclear physics. On the other hand, Bethe [13], Bethe and Goldstone [14] and Rodberg [15] introduced the concepts of "wave" and "reaction" operators to extend this method to study the nuclear many-body problem. Goldstone [8] derived the completely linked (to all order) diagrammatic MBPT using the field theoretic form of the time-dependent perturbation theory and adiabatic hypothesis [16]. Hugenholtz [9] arrived at the same conclusion via a time-independent resolvent approach, leading to the emergence of the famous "linked cluster theorem" (LCT) for closed-shell systems. An early attempt on a degenerate case was made by Morita [17]. The first complete proof of this theorem for a degenerate situation was provided by Brandow [18] using the time-independent approach and later by Lindgren [19].

The closed-shell coupled cluster (CC) method, which is an all-order version of the MBPT, was first introduced by Coester[20] and Coester and Kümmel [21] in nuclear physics. Coester and Kümmel [21] pioneered the use of the exponential ansatz in formulating a quantum mechanical many-body theory in the realm of nuclear physics, although the origin of the cluster expansion approach to many-fermion systems dates back to the early 1950s through attempts to understand the correlation effect in nuclear matter and electron gas by Bethe [13] and Gell-Mann and Brueckner [22]. At about the same time, Cížek [23] developed the formalism of the CC approach for use in the context of molecular electronic structure theory. With the advent of computing methodologies and algorithms during the 1970s, both CC and MBPT were implemented numerically for molecular systems having a reasonably high degree of complexity. Applications of the ab initio CC theory using finite basis set expansions were put forth in 1972 by Paldus, Čížek and Shavitt [24], and in the following year Kaldor [25] came up with his application of the MBPT to the hydrogen molecule ground state by invoking the algebraic approximation. The development of high-speed computers, the refinement of algorithmic techniques, and a plethora of fundamental formal advancements during the 1970s and early 1980s culminated in a rapid and vivid understanding of electron correlation problems in atomic and molecular systems.

1.2 Born–Oppenheimer approximation

It is well known that particles at the atomic or subatomic scale, like electrons and the nucleus in atoms and molecules, must be treated quantum mechanically, as Newtonian mechanics fails to provide the correct description of the behavior of these particles. The quantum mechanical methodology finds one of its most important domains of application in the evaluation of atomic and molecular properties. The method is nowadays applied to almost all branches of physics and chemistry. It also finds its applications in atomic and nuclear physics to characterize spectral lines and nuclear structure.

In this methodology, a wave function is needed to explain the kinematics of electrons in atoms and molecules, where the wave function Ψ defines the quantum state of a system. The wave function Ψ generally depends on time and has the physical interpretation that $|\Psi(\zeta, t)|^2$ represents the probability density of finding the system at position ζ at the instant of time t. The quantum state Ψ is governed by the Schrödinger equation, which is the first-order differential equation in t of the form

$$i\hbar \frac{d\Psi(\zeta, t)}{dt} = \mathcal{H}\Psi(\zeta, t) \tag{1.1}$$

where \mathcal{H} is the Hamiltonian of the system. The fundamental task is to partition Equation (1.1) into space (ζ) and time (t) and solve the time-independent part of the Schrödinger equation

$$\mathcal{H}|\Psi(r,R)\rangle = \mathcal{E}|\Psi(r,R)\rangle \tag{1.2}$$

In atomic units, the Hamiltonian \mathcal{H} for N electrons and M nuclei is given by

$$\mathcal{H} = -\frac{1}{2} \sum_{i=1}^{N} \nabla_{i}^{2} - \sum_{A=1}^{M} \frac{1}{2M_{A}} \nabla_{A}^{2} - \sum_{i=1}^{N} \sum_{A=1}^{M} \frac{Z_{A}}{r_{iA}} + \sum_{i=1}^{N} \sum_{i(1.3)$$

in which M_A is the ratio of the mass of nucleus A to the electron mass. The first two terms in the above equation represent the kinetic energies of the electron and the nucleus. The third term refers to the electrostatic interaction between the nucleus and the electrons, while the last two terms denote the electronelectron and nucleus-nucleus repulsions. This Hamiltonian is universal in the sense that it describes all of everyday matter from biological macro molecules, such as proteins, enzymes and nucleic acids, to metals and semiconductors to synthetic materials such as plastics. Thus, if we could solve for the eigenvalues and eigenfunctions of this Hamiltonian, we could predict any property we wished of a given system. This fact led the physicist P. A. M. Dirac to comment that all of chemistry is a solved problem, at least in principle. Of course, the problem cannot actually be solved exactly, so approximations are needed and it is the development of reliable approximate methods on which quantum chemistry is focused.

Electrons, being lighter than the nuclei, move faster than the nuclei, and hence, to a good approximation one can assume the nuclei to be stationary. The large separation in the time scales of electronic and nuclear motions is rationalized by the fact that the energy separation between the electronic energy levels is much larger than the rotational and/or the vibrational energy levels. This is akin to a situation where one might contemplate that the electrons are being dragged along with the nuclei. From a dynamical perspective, this is equivalent to stating that electrons follow the motion of the nuclei in an adiabatic manner. This is the so-called Born–Oppenheimer (BO) approximation [26, 27]. In this approximation, the nuclear kinetic energy terms are neglected and the nuclear repulsion energy terms are considered to be constant. The reason it is an approximation is because, in principle, one cannot ignore the role of the non-adiabatic effects such as spontaneous and instantaneous motion of the electrons along with the nuclei. Nevertheless, in the case of many-electron systems, the adiabatic approximation is an excellent idea to work with. The mass difference between the nuclei and the electrons in molecules generates yet another effect. The nuclear components of the wave function are seen to be much more localized, spatially, in comparison to electronic wave functions (making the nuclear wave functions rise more steeply than the corresponding electronic counterparts). This effect, in the classical limit, allows one to treat the nuclei as classical point masses. Thus, the electronic Hamiltonian or the Hamiltonian describing the motion of N electrons in a field of M point charges can be expressed as

$$\mathcal{H}_e = -\frac{1}{2} \sum_{i=1}^{N} \nabla_i^2 - \sum_{i=1}^{N} \sum_{A=1}^{M} \frac{Z_A}{r_{iA}} + \sum_{i=1}^{N} \sum_{i(1.4)$$

Assuming the eigenfunction of the Hamiltonian \mathcal{H} to be product separable

$$\Psi(r, R) = \Phi_e(r, \{R\}) \Phi_n(R)$$
(1.5)

the Schrödinger equation for the electronic Hamiltonian can be written as

$$\mathcal{H}_e \Phi_e(r, \{R\}) = \mathcal{E}_{\text{elec}}(r, \{R\}) \Phi_e(r, \{R\})$$
(1.6)

in which the parametric dependence of the eigenvalue $\mathcal{E}_{elec}(r, \{R\})$ and the eigenfunction $\Phi_e(r, \{R\})$ on the nuclear coordinate is explicitly acknowledged. The total energy of the system at a given geometry can then be obtained by